

ISOCONCINNDIOL, A BROMINATED DITERPENOID FROM *LAURENCIA SNYDERAE* VAR. *GUADALUPENSIS*

BRUCE M. HOWARD* and WILLIAM FENICAL†

* Department of Chemistry, San Francisco State University, San Francisco, CA 94132, U.S.A.; † Institute of Marine Resources, Scripps Institution of Oceanography, La Jolla, CA 92093, U.S.A.

(Revised received 11 April 1980)

Key Word Index *Laurencia snyderae* var. *guadalupensis*; Rhodomelaceae; marine natural products; isoconcinndiol.

Abstract—A new brominated diterpenoid, isoconcinndiol, has been isolated from the marine red alga *Laurencia snyderae*.

INTRODUCTION

Previous investigations of the natural products chemistry of the marine alga *Laurencia snyderae* Dawson (Rhodomelaceae, Rhodophyta) have resulted in the isolation of the brominated diterpenoid concinndiol (1) [1]. Concinndiol was first isolated as a natural product from *Laurencia concinna* from Australia [2]. In our studies of the natural products chemistry of various unknown *Laurencia* species from the Galapagos Islands, we [3] have isolated the closely related brominated diterpenoid aplysin-20 (2) which was first described as a constituent of the Japanese sea hare, *Aplysia kurodai* [4]. More recently a tricyclic representative of this class, isoplysin-20 (3), was also isolated from *Aplysia* [5]. We wish to report here the isolation and structure elucidation of a new diterpenoid of this class, isoconcinndiol (4).

RESULTS AND DISCUSSION

Open column silica gel chromatography of the chloroform-methanol extract of fresh *Laurencia snyderae* var. *guadalupensis* Dawson (Guadalupe Island, Mexico, 1978) gave, upon diethyl ether elution, isoconcinndiol (4), as a white crystalline solid, mp 173-174°. Mass spectral analysis of 4 established a molecular formula of $C_{20}H_{33}BrO$ for the $M^+ - H_2O$ fragment at m/e 368/370, and showed fragments at m/e 270/272 and 191 (Scheme 1) which are identical with the mass spectral behavior of both 1 and 2. The infra-red spectrum showed hydroxyl (3500 cm^{-1}) and olefinic absorptions ($3000, 999$ and 920 cm^{-1}). The 1H NMR spectrum of 4 revealed the presence of a terminal vinyl group [δ 5.91 (1H, dd, $J = 15, 10\text{ Hz}$), 5.17 (1H, dd, $J = 15, 2\text{ Hz}$) and 5.02 (1H, dd, $J = 10, 2\text{ Hz}$)], an alpha to bromine proton with bromine situated equatorial in a cyclohexane ring [δ 3.95 (1H, dd, $J = 12, 4\text{ Hz}$)], and five methyl singlets, which are similar features exhibited by compounds 1 and 2.

Several attempts were made to interconvert 1, 2 and 4. Methods involving acid-catalysed allylic hydroxyl rearrangements gave only mixtures of inseparable brominated hydrocarbons. Treatment of 2 and 4 with pyridinium chlorochromate gave the same α, β -unsaturated aldehyde as determined by TLC. However,

this aldehyde proved to be unstable and could not be isolated in pure form.

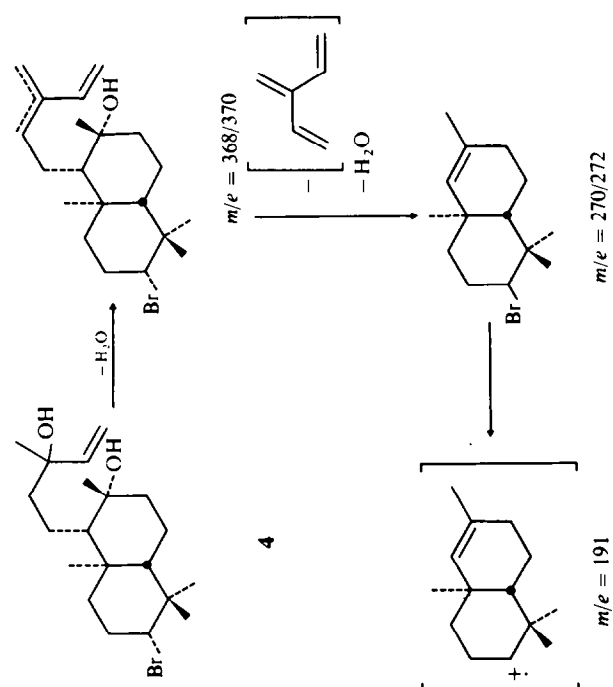
To place the structure proposal for isoconcinndiol on a more firm basis, a detailed comparison of the ^{13}C NMR spectra of compounds 1, 2 and 4 was performed. Table 1 presents the ^{13}C NMR spectra of these brominated diterpenoids in which all methine and quaternary carbons are assigned. These assignments were greatly aided by off-resonance proton decoupling. The carbons which structurally differ between 1, 2 and 4 are C-8, C-9, C-13, C-14 and C-15. However, isoconcinndiol (4) differs from concinndiol (1) only at C-8 and C-9 and aplysin-20 (2)

Table 1. ^{13}C NMR chemical shift assignments for concinndiol (1), aplysin-20 (2) and isoconcinndiol (4)*

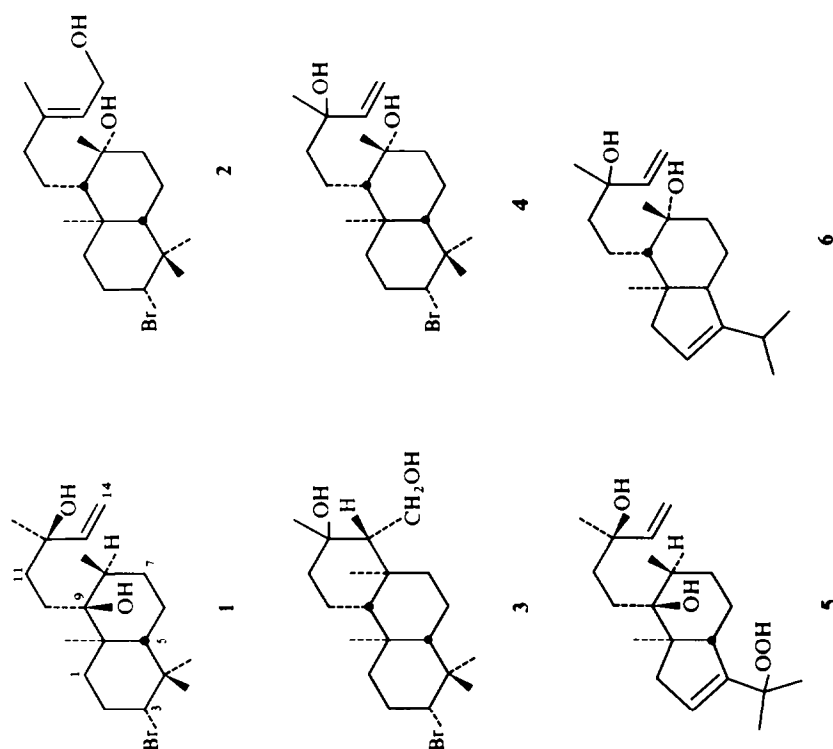
Carbon	1	2	4
3	71.2 d	70.9 d	70.6 d
4	40.2 s	40.4 s	40.0 s
5	47.5 d	57.0 d	57.1 d
8	38.3 d	72.2 s	70.6 s
9	76.8 s	59.1 d	62.3 d
10	44.2 s	39.8 s	41.7 d
13	73.2 s	138.1 s	73.7 d
14	147.0 d	125.7 d	147.2 d
15	111.4 t	59.5 t	111.2 t
<i>Unassigned Methyl and Methylene Carbons</i>			
	37.1	44.0	45.9
	34.5	43.3	45.0
	32.7	41.3	32.7
	32.1	31.7	30.8
	†	31.6	30.6
	†	30.7	29.3
	†	30.7	24.4
	24.0	24.7	22.7
	18.7	20.7	20.0
	16.8	18.8	18.5
	16.7	15.7	15.8

* All spectra run in acetone- d_6 solution and reported in ppm relative to TMS(0).

† Signals obscured by acetone- d_6 .



Scheme 1. Mass spectral fragmentation of isoconcinidiol.



differs from isoconcinndiol only at C-13, C-14 and C-15. These relationships are substantiated in their respective ^{13}C NMR spectra.

Evidence for the presence of a C-8 axial hydroxyl in isoconcinndiol as in **2** rather than a C-8 equatorial hydroxyl as in sclareol was provided by mass spectral analysis. Sclareol and similar diterpenes with a C-8 equatorial hydroxyl or an exocyclic double bond show a prominent fragmentation between C-6 and C-7 and C-9 and C-10 (an $\text{M}^+ - 170$ for sclareol) [6]. This fragmentation is absent in both **2** and **4** and suggests that the C-8 hydroxyl on **4** is axial.

It has been suggested that nonhalogenated terpenes in *Laurencia* may arise from halogenated precursors [7]. For instance, neoconcinndiol hydroperoxide (**5**) may be envisioned to be derived from concinndiol (**1**) via solvolysis of the C-3 equatorial bromine followed by concerted ring contraction, elimination and subsequent oxidation [3]. In an effort to examine the rearrangement potential of brominated labdanes, **4** was subjected to reaction with various silver salts. Treatment with AgBF_4 in anhydrous diethyl ether at -20° for 3 hr gave a smooth conversion to the ring-contracted product **6** (77% yield). The facile nature of this reaction lends support to these previous contentions.

EXPERIMENTAL

Isolation of isoconcinndiol (4). Freshly collected *Laurencia snyderae* var. *guadalupensis* (10 kg) was homogenized with MeOH and the homogenate repeatedly extracted with CHCl_3 to yield 60 g of crude extract after solvent removal *in vacuo*. Fractionation over Si gel (Grace Chemical, grade 62) gave, upon Et_2O elution, fractions which yielded crystalline samples of **3**. Recrystallization from CHCl_3 gave analytically pure isoconcinndiol, mp $173-174^\circ$. ^1H NMR (220 MHz, CDCl_3): δ 5.95 (1H, dd, $J = 15, 10$ Hz), 5.17 (1H, dd, $J = 15, 2$ Hz), 5.02 (1H, dd, $J = 10, 2$ Hz), 3.95 (1H, dd, $J = 12, 4$ Hz), 3.10 (1H, s, OH), 2.42 (1H, s, OH), 1.24 (3H, s), 1.14 (3H, s), 1.06 (3H, s), 0.90 (3H, s), 0.81 (3H, s); IR $\nu_{\text{max}}^{\text{KBr}}$: 3500, 3050, 2950, 1450, 1390, 1370, 1315, 1185, 1160, 999, 920; MS (70 eV) m/e : 368; 370 ($\text{M}^+ - \text{H}_2\text{O}$), 270; 272 ($\text{M}^+ - 2\text{H}_2\text{O} - \text{C}_6\text{H}_8$), 191 ($\text{M}^+ - 2\text{H}_2\text{O} - \text{C}_6\text{H}_8 - \text{Br}$).

Attempted oxidation of aplysin-20 (2) and isoconcinndiol (4) with pyridinium chlorochromate. In separate

experiments 30 mg of compounds **2** and **3** were dissolved in CH_2Cl_2 (5 ml) and excess pyridinium chlorochromate (100 mg) was added with stirring. After 1 hr TLC analyses (petrol- Et_2O , 80:20) of the reaction mixtures showed that identical UV active aldehydes had been produced. The reaction mixtures were passed over a Si gel column for work-up and the CH_2Cl_2 removed *in vacuo*. However, the aldehyde produced proved to be unstable to these work-up conditions and could not be isolated in pure form.

Silver tetrafluoroborate rearrangement of isoconcinndiol. Isoconcinndiol (**4**) (50 mg, 0.13 mmol) was dissolved in dry Et_2O (20 ml) and cooled to -20° with an ice Me_2CO bath. AgBF_4 (125 mg, 0.65 mmol; 5-fold excess) was added and the suspension was stirred at -20° . After 3 hr the reaction mixture was poured onto ice and extracted with Et_2O (3×50 ml). Prep. TLC of the reaction mixture (Si gel, Et_2O -hexane 1:1) gave pure samples of the rearrangement product, **6** (30 mg, 0.10 mmol, 77% yield) as an oil. ^1H NMR (60 MHz, CCl_4): δ 5.83 (1H, dd, $J = 17, 10$ Hz), 5.13 (1H, dd, $J = 17, 2$ Hz), 4.90 (1H, dd, $J = 10, 2$ Hz), 1.28 (3H, s), 1.25 (3H, s), 0.98 (3H, d, $J = 7$ Hz), 0.92 (3H, d, $J = 7$ Hz), 0.85 (3H, s).

Acknowledgements We wish to thank the captain and crew of R/V Ellen B. Scripps for logistical support in collecting *L. snyderae* var. *guadalupensis* at Guadalupe Island, Mexico. This work is a result of research funding provided to the Scripps Institution by NOAA, Office of Sea Grant, Department of Commerce, under Grant No. 04-7-158-44121. The U.S. Government is authorized to produce and distribute reprints for governmental purposes, notwithstanding any copyright notation that may appear hereon.

REFERENCES

- Howard, B. M. (1978). Ph.D. Thesis, University of California, San Diego.
- Sims, J. J., Lin, G. Y., Wing, R. M. and Fenical, W. (1973). *Chem. Commun.* 470.
- Howard, B. M., Fenical, W., Finer, J. and Clardy, J. (1977). *J. Am. Chem. Soc.* **99**, 6440.
- Yamamura, S. and Hirata, Y. (1971) *Bull. Chem. Soc. Jpn.* **44**, 2560.
- Yamamura, S. and Terada, Y. (1977) *Tetrahedron Letters* 2171.
- Enzell, C. (1961) *Acta Chem. Scand.* **15**, 1303.
- Fenical, W. (1979) in *Recent Advances in Phytochemistry*, Volume 13. (Swain, T. and Waller, G. W., eds.) pp. 219-239. Plenum Press, New York.